

Design, synthesis, and SAR studies of novel and highly active tri-cyclic HIV integrase inhibitors

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Abstract—A novel class of tri-cyclic HIV integrase inhibitors were designed based on conformational analysis of 1,6-naphthyridine carboxamide compound L-870810 and docking the designed inhibitor into the active site of our integrase enzyme model. The efficient syntheses of pyrroloquinoline tri-cyclic analogs are described. The SAR studies resulted in the identification of a lead compound that is more potent and more soluble than L-870810.

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Many marketed anti-HIV therapeutic agents target either of the two viral enzymes, HIV reverse transcriptase or HIV protease. However, because of continuously emerging HIV mutants that are resistant to most of these drugs, there is the need for HIV inhibitors with a novel mechanism of action.¹ One of the key steps in the life cycle of HIV is the integration of a pro-viral DNA into the genome of the host cell in the nucleus. This crucial event is catalyzed by a third virally encoded and packaged enzyme, HIV integrase. The integration process involves multiple steps and factors from both the virus and the host.² The recent clinical trials of an HIV integrase inhibitor L-870810 (**1**) by Merck showed encouraging results of HIV viral load reduction in AIDS patients,³ thus validating integrase as an attractive HIV therapeutic target.

As depicted in Figure 1, there are two rotational isomers associated with the amide bond in **1**. While conformer B has been postulated as the active form for binding,^{3b} conformer A is the lower energy species in the free form based on our calculations.⁴ Although there is a difference of only 1.7 kcal/mol between A and B, conversion from A to B will cost approximately 5 kcal/mol to overcome the rotational barrier.

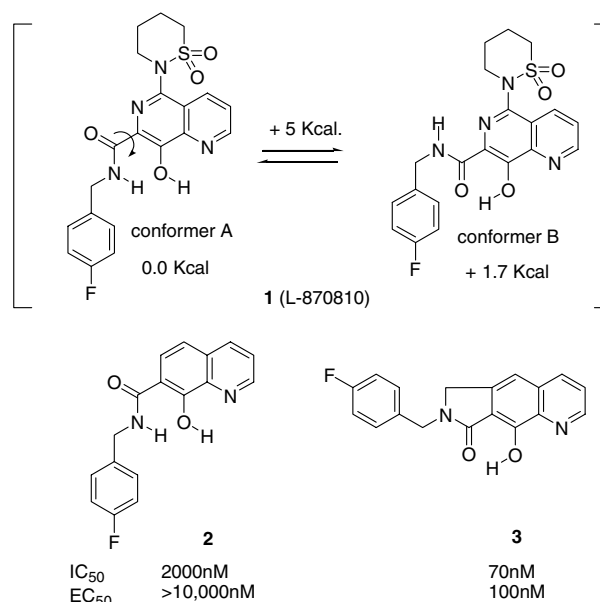


Figure 1. The structures of L-870810 (**1**), quinoline carboxamide (**2**), and tri-cyclic pyrroloquinoline (**3**). IC₅₀: the concentration required to inhibit 50% of integrase strand transfer activity; EC₅₀: anti-HIV activity in MT-4 cells; for details, see Ref. 9a) and 9b).

Hence, it was hypothesized that a pre-organized scaffold such as **3** would provide an energetic benefit for binding in the active site. To directly evaluate the effects of ring

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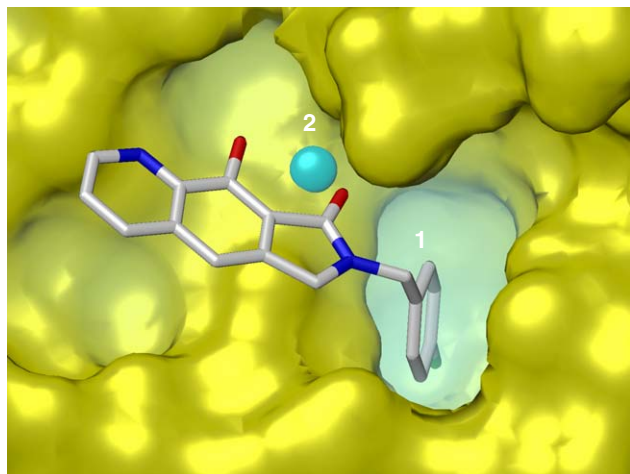


Figure 2. Top view of compound **3** docked into the active site of an integrase/DNA complex model.

closure on the activity, we examined a pair of prototype compounds **2** and **3** shown in Figure 1.⁵

As expected, compound **3** is significantly more potent than compound **2**. The results are consistent with our modeling which docked compound **3** into the active site (Fig. 2) of an integrase/DNA complex model.⁶ In the model, a hydrophobic pocket (labeled 1) exists and is well utilized by the *p*-fluorobenzyl moiety that is held tightly in the optimal direction by the lactam ring. This discovery prompted us to initiate a program to investigate several series of HIV integrase inhibitors designed on the pre-organized tri-cyclic core.^{7,8} Our active site model also depicts specific polar interactions between inhibitors and enzyme⁶ in addition to the crit-

ical interaction between the magnesium and the inhibitor (labeled 2).

Our synthetic strategies were planned based upon (a) quick assembly of a tri-cyclic scaffold and (b) the ability to create opportunities for optimizing both the potency and other pharmacological properties of analogs. The synthesis of the tri-cyclic core **5** and two immediate analogs is depicted in Scheme 1 starting with the Dieckmann condensation between **4**, prepared easily from succinimide, and the commercially available pyridine-2,3-dicarboxylic acid dimethyl ester. The differentiation

Table 1. Integration strand transfer inhibition, anti-HIV proliferation and cytotoxicity assay results for compounds **8–10**, **14**, and **16–19**^a

Compound	IC ₅₀ ^b	EC ₅₀ ^c (MT-4 cell)	CC ₅₀ ^d (MT-4 cell)	Solubility ^e
8	0.08	7.5	19	<1.4
9	2.4	1.18	22	2.5
10	0.05	0.089	5.1	3
14	0.76	0.003	1.5	nd
16	1.2	0.78	1	nd
17	2.1	0.34	1	nd
18	0.35	0.07	0.4	nd
19	0.335	0.011	0.3	nd

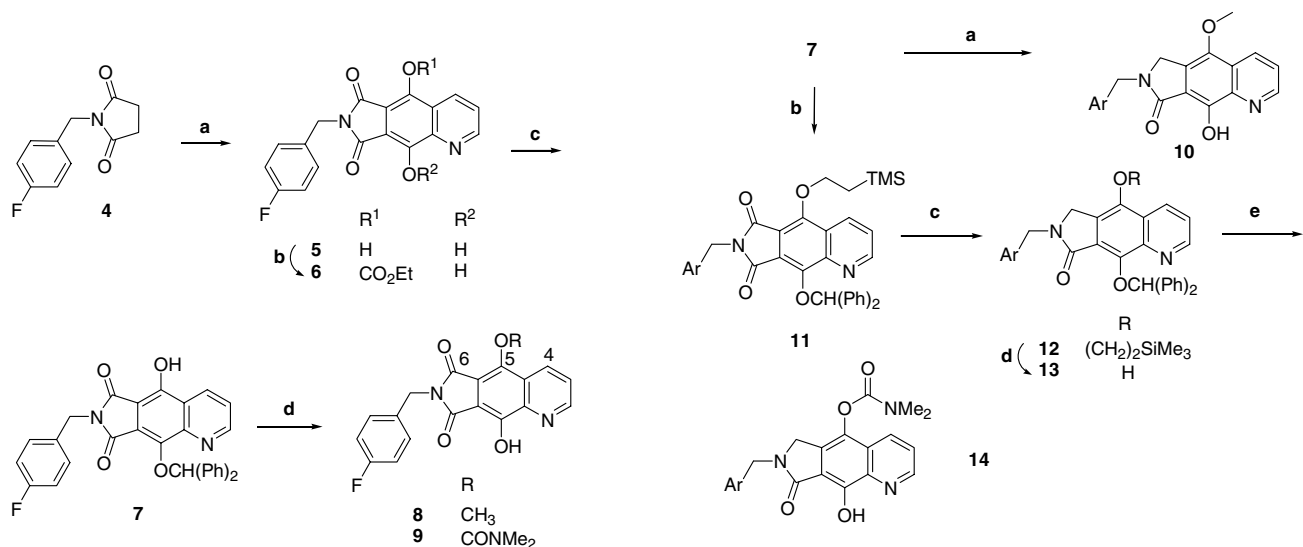
^a Values are means of at least two experiments, given in μ M, nd: not determined.

^b Ref. 9a.

^c Ref. 9b.

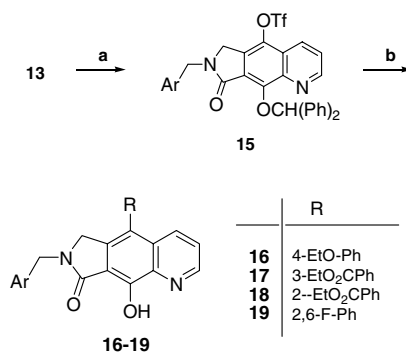
^d Ref. 9c.

^e Measured by dissolving the solid of testing compound in 20% of acetonitrile or 20% of ethanol in phosphate buffer (pH 7.3) and determining the concentration of supernatants by HPLC after incubation at rt overnight.

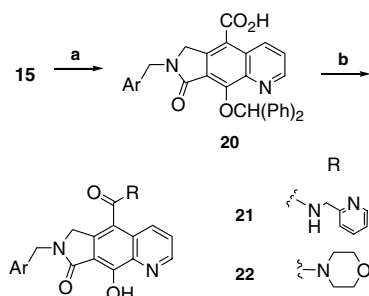


Scheme 1. Reagents and conditions: (a) NaOMe, THF, pyridine-2,3-dicarboxylic acid dimethyl ester, reflux, 24 h, 66%; (b) ethyl chloroformate, NaOH, 1,4-dioxane/water, rt, 1 h, 63%; (c) 1-(Ph)₂CN₂, 1,2-dichloroethane, 70 °C, 3 h, 80%; 2—K₂CO₃, THF/water, cat. 4-dimethylaminopyridine, rt, 12 h, 100%; (d) 1—for **8**: iodomethane, K₂CO₃, DMF, 40 °C, 1 h, 85%, for **9**: ClCONMe₂, TEA, DMAP, dichloromethane, rt, 12 h, 50%; 2—trifluoroacetic acid, triethylsilane, rt, 10–30 min, 64–68%.

Scheme 2. Reagents and conditions: Ar: *p*-fluorophenyl; (a) 1—MeI, K₂CO₃, DMF, 40 °C, 1 h; 2—NaBH₄, THF/MeOH, rt, 15 h, 3—triethylsilane, TFA, 80% for three steps; (b) 2-trimethylsilylethanol, DEAD, Ph₃P, dichloromethane, rt, 3 h, 82%; (c) 1—LiBH₄, THF/water, 0 °C to rt, 1 h; 2—Ac₂O, DMAP, DCM, rt, 12 h, 3—triethylsilane, TMSOTf, 2,6-lutidine, rt, 12 h, 47% for three steps; (d) TBAF, THF, rt, 1 hr, 100%; (e) 1—dichloromethane, ClCONMe₂, DMAP, rt, 4 h, 47%; 2—trifluoroacetic acid, triethylsilane, rt, 30 min, 93%.



Scheme 3. Reagents and conditions: Ar: *p*-fluorophenyl; (a) Cs₂CO₃, *N*-phenyltrifluoromethane sulfonamide, dichloromethane, reflux, 1 h, 43%; (b) (16–19) 1–4-ethoxyphenylboronic acid (for 16); 3-ethoxycarbonylphenylboronic acid (for 17); 2-ethoxycarbonylphenylboronic acid (for 18) or 2,6-difluorophenylboronic acid (for 19), tetrakis-(triphenylphosphine)-palladium(0), K₂CO₃, toluene, ethanol, water, 120 °C; 3 h, 17–48%, 2—triethylsilane, TFA, dichloromethane, rt, 1 h, 30–68%.



Scheme 4. Reagents and conditions: Ar: *p*-fluorophenyl; (a) 1—Pd(OAc)₂, dppp, TEA, DMF/water, CO (1 atm), 70 °C, 2.5 h; 2—Cs₂CO₃, iodomethane, rt, 45 min, 70%, two steps; 3—LiOH, THF/water, rt, 15 h, 98%; (b) 1—2-aminomethylpyridine (for 21) or morpholine (for 22), diisopropylethylamine, HATU, DMF, rt, 15 h, 2—triethylsilane, TFA, dichloromethane, rt, 1 h, 30–53% for two steps.

of the two phenolic hydroxyl groups was accomplished using the Schotten–Baumann protocol to form mono-ethyl carbonate 6. The free hydroxyl was then protected as a diphenylmethyl (DPM) ether, which possesses both stability to basic conditions and desirable steric hindrance to facilitate the control of regioselectivity in the subsequent modification of the succinimide moiety.

Upon standard hydrolysis, compound 7 was converted to 8 and 9. Their activities are reported in Table 1.

Succinimide analog 8 lacked significant anti-HIV potency in the cell-based assay, while showing good integrase inhibitory activity. Similar structures were reported by researchers from the Tibotec.¹⁰ Encouraged by the activity of 8 in the strand transfer assay, we undertook further modification by removing the C6 carbonyl to improve the potency in the cell-based assay. As shown in Scheme 2, compound 7 was converted to the corresponding methyl ether 10 and carbamate 14.

Compared to 8 and 9, compounds 10 and 14 exhibited substantially improved potency in the cell-based assay (Table 1). The improved potency of 10 over 8 could be due to the increased solubility and permeation through the cell membrane. The enhanced activity of C5 carbamate 14 in the cell-based assay as compared to C5 methyl ether 10 was also noticeable. Therefore, we explored other functional groups at C5 in addition to the carbamate. Particularly, compounds containing the C5 substituents linked with a C–C bond were investigated because of their stability. Schemes 3 and 4 depict the syntheses of selected bi-aryl and carboxamide analogs.

Compound 18 is the most active among the three simple mono-substituted bi-aryl analogs 16–18 (Table 1). The potency in the HIV cell-based assay was further enhanced when a 2,6-difluorophenyl group was introduced in compound 19. It is unclear what factors contributed to the potency improvement in this series of C5 bi-aryl analogs.

Both 21 and 22¹¹ displayed IC₅₀ in sub-micromolar and EC₅₀ in low nanomolar ranges (Table 2). We also tested these compounds for the protein-shifted anti-HIV activity. When taking into account the binding effects by both human serum albumin and α-1-acid glycoprotein, these compounds showed potency similar to or better than 1. A plausible explanation for the reduced effect of these proteins on the potency of 22 as compared to 1 is the higher polarity of 22, as shown by the calculated log *P*. We also measured the solubility of compounds 1, 21, and 22 at pH 7.3 and found that both 21 and 22 are more soluble than 1 in water.¹² These data point to the potential of tri-cyclic analogs as promising candidates for further pharmacological evaluation.

Table 2. Activities of carboxamide analogs and selected data of protein binding effect on potency, log *P* and solubility^a

Compound	IC ₅₀ ^b	EC ₅₀ ^c (MT-2 cell)	CC ₅₀ ^d (MT-4 cell)	Protein-shifted EC ₅₀ ^e (MT-2 cell)	Calcd log <i>P</i> ^f	Solubility ^g
21	0.19	0.0098	1.6	0.094	2.12	36
22	0.036	0.0034	4	0.054	0.97	97
1	0.05	0.0038	1.5	0.088	2.24	7.2

^a All values are given in μM except for calcd log *P*.

^b Ref. 9a.

^c Ref. 9b.

^d Ref. 9b.

^e Ref. 9c.

^f The program used for calculations: Pallas.

^g Measured by dissolving the solid of testing compound in 20% of acetonitrile or 20% of ethanol in phosphate buffer (pH 7.3) and determining the concentration of supernatants by HPLC after incubation at rt overnight.

The synthetic schemes described above have allowed us to examine a variety of structures based on a pre-organized pyrroloquinoline tri-cyclic core and evaluate their inhibitory activities against HIV integration in both the strand transfer and cytopathic anti-HIV assays. We have discovered novel, highly organized and potent HIV integrase inhibitors from this class of compounds. Furthermore, our lead compound **22** is more soluble and more potent than the clinically efficacious compound **1** when tested under the physiologically relevant conditions. The identification of a compound demonstrating in vivo pharmacological properties for further development as a clinically useful HIV inhibitor is the subject of our ongoing research program and results will be published in due course.

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- See Refs. **9a** for IC₅₀ and **9b** (MT-4 cell) for EC₅₀. Compound **2** was prepared from commercially available 8-hydroxyl-quinoline-7-carboxylic acid and 4-fluorobenzylamine by standard amide-forming conditions similar to reaction b in **Scheme 4**. Compound **3** was obtained from triflate **15** in **Scheme 3** according to condition b.
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